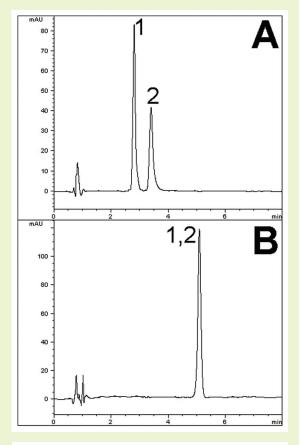
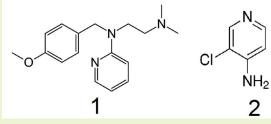


Pyrilamine and 4-Amino-3-Chloropyridine

Unique selectivity from a unique stationary phase





Note: Amine-containing compounds such as pyrilamine and 4-amino-3-chloropyridine can be difficult to analyze using conventional silicabased stationary phases. These columns have residual silanol groups on the surface that can interact electrostatically with amines, causing peak tailing. Chromatographers use a number of strategies to avoid these issues, such as use of ion pair agents or endcapping. However, Cogent TYPE-C Silica phases are virtually free of silanols, and therefore good peak shapes can be obtained without these workaround method strategies.

Method Conditions

Column: Cogent Amide™, 4µm, 100Å

Catalog No.: 40036-05P

Dimensions: 4.6 x 50 mm

Solvents: A: 90% DI H₂O / 10% acetonitrile / 0.1% formic acid (v/v) B: Acetonitrile / 0.1% formic acid (v/v)

Gradient:	time (min.)	%B
	0	90
	1	90
	7	50
	8	90

Post Time: 3 min

Injection vol.: 2µL

Flow rate: 1.0mL/min

Detection: UV 244 nm

Sample: 100 mg/L pyrilamine and 4-amino-3-chloropyridine reference standards in diluent of 50/50 solvent A/solvent B. Peak identities confirmed with individual standards

Peaks: 1. Pyrilamine 2. 4-Amino-3-chloropyridine

Discussion

The Cogent Amide column offers unique selectivity that may not be readily attainable with other phases. Two test solutes shown in this application note (pyrilamine and 4-amino-3-chloropyridine) were baseline separated on the Cogent Amide column (Figure A), but they co-eluted with no resolution on a different Cogent column using otherwise equivalent method conditions (Cogent Diamond Hydride[™], Figure B). The presence of the amide ligand provides additional selectivity that can make a significant difference in resolving closelyeluting compounds such as these.

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