

## Vitamin K Isomers by Reverse Phase

Phytonadione separation by shape selectivity

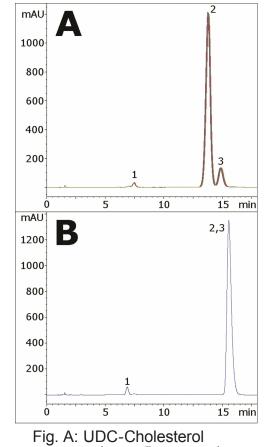
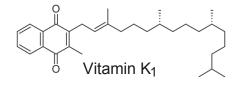


Fig. B: Ordinary C8 column



**Note:** Phytonadione (a.k.a. Phylloquinone, Vitamin  $K_1$ ) is a lipophilic vitamin that can be obtained in the diet from leafy green vegetables. It plays an essential role in blood clotting by acting as a cofactor for formation of coagulation factors II, VII, IX, and X. The letter designation for Vitamin K was based on the first letter of "Koagulationsvitamin" (coagulation vitamin), which is from the German journal that first published its identification by Danish biochemist Henrik Dam.

## **Method Conditions**

Column: Cogent UDC-Cholesterol™, 4µm, 100Å

Catalog No.: 69069-15P

Dimensions: 4.6 x 150 mm

Solvents: A: 50% DI H<sub>2</sub>O / 50% MeOH / 0.1% formic acid B: 97% acetonitrile / 3% DI H<sub>2</sub>O / 0.1% formic acid

Gradient:	time (min.)	%B
	0	80
	15	92
	16	80

Temperature: 12°C

Post time: 2 min

Flow rate: 1.5 mL/min

Detection: UV 254 nm

 Sample: Stock Solution: 10μL/mL phytonadione in acetonitrile diluent. (The solution was vortexed for 10 minutes.)
Working Solution: Stock solution was diluted 1:10 with acetonitrile.

Peaks: 1. Impurity

2. Phytonadione (E isomer)

3. Phytonadione (Z isomer)

**t<sub>0</sub>:** 1.0 min

## Discussion

Using the USP methodology, assay of Vitamin K<sub>1</sub> (phytonadione) is done in organic normal phase (ONP) with a bare silica column. This can be inconvenient for many laboratories which may not have a dedicated ONP instrument and therefore must spend both time and solvents to convert their reverse phase system to ONP for the analysis. In reverse phase, adequate separation between the E and Z isomers of phytonadione may not be obtained using conventional alkyl chain-based stationary phases (as Figure B illustrates). However, a resolution of 1.5 was obtained between isomers using the Cogent UDC-Cholesterol column (shown in Figure A), which meets the USP requirement for resolution. The Cogent UDC-Cholesterol is able to separate the two isomers on the basis of shape selectivity.

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