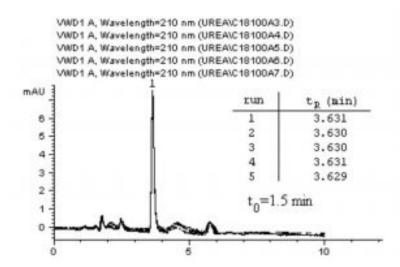


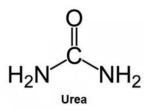
## Urea Analyzed by HPLC- AppNote

## Fast, Reproducible Method

In this Method Urea was retained when a 100% DI Water Mobile Phase was used. DI Water can be used as the complete Mobile Phase without fear of phase "de-wetting" and subsequent loss of retention time.

This Method has the advantage of being very simple, without time consuming derivatization or reaction steps when compared to other methods. Urea can be determined in complex mixtures using this fast, reproducible (see insert on the chromatogram) Method which could also be useful in the analysis of biological samples for diagnostic purposes.





Peak:

Urea

## **Method Conditions**

Column: Cogent Bidentate C18<sup>™</sup>, 4µm, 100Å

Catalog No.: 40018-15P

Dimensions: 4.6 x 150mm

Detection: UV @ 210nm

**Mobile Phase:** DI Water (Isocratic run: 100% A)

Flow rate: 0.5mL/minute Copyright 2024, All Rights Apply

Injection vol.: 10µL 9158 Industrial Blvd. NE, Leland, NC 28451

Sample Preparation: 1mg of Urea standard was added to 1ml of DI Water. tel. (732) 380-8900, fax (910) 769-9435

Email: customers@mtc-usa.com

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MicroSolv Technology Corporation

Website: www.mtc-usa.com



Note: There is growing demand for a sensitive and reliable procedure for the determination of Urea in many matrices such as milk, soil extracts, seawater and wine. In addition there are several clinical applications for the analysis of this compound. The most common approach for measurement of Urea involves detection of ammonia (after hydrolysis) by color forming reactions – enzymatic, colorimetric Methods. The other techniques require noxious reagents and produce an unpleasant odor [1]. Newer Methods involve high-performance thin layer chromatography-densitometry, alkalimetric titration. HPLC is the most specific method but either organic Normal Phase Chromatography or Ion Pairing Chromatography have to be used to retain this very polar compound until this Method.

[1] "Determination of urea using HPLC with fluorescence detection after automated derivatization with xanthydrol", S. Clark, P.S. Francis, X.A. Conlan, N.W. Barnett, J. Chromatography A, 1161 (2007) 207-213.



## Attachment

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